

# THE DETERMINATION OF TRACE ELEMENTS IN SQUIDS USING ICP-MS/INAA AND ITS APPLICATION TO THE MARINE RADIOACTIVITY MONITORING

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## 1. INTRODUCTION

The atmospheric nuclear testing of nuclear weapon conducted in the late 1950s and 1960s increased the level of environmental radioactivity in the ecosystem. Moreover, since the accident of Chernobyl nuclear power plant in 1986, radioactive fallout had been world-widely spreaded out, and much more attention has been directed to monitoring the environmental radioactivities in the ecosystem and in the food in aspect of radiological hazard. As the number of nuclear power plants in many countries were continuously increased in the recent years, there might exist the discharge of a large amount of radionuclides into environments and of their environmental contamination.

In the marine ecosystem, some marine organisms such as algae and shellfish are the first concentrators of metals or radionuclides and, through the food chain, those substances are transferred to fishes and finally human being. It is well accepted that the analysis of trace elements in

marine organisms is very important for evaluating the environmental contamination of radionuclides as well as estimating their behaviors in the marine ecosystem.

In 1965, Folsom and Young found  $^{60}\text{Co}$  and  $^{108\text{m}/110\text{m}}\text{Ag}$  in the viscera of squid collected from the North Pacific. After his discover, we had continuously surveyed the long lived radionuclides including  $^{60}\text{Co}$  and  $^{108\text{m}/110\text{m}}\text{Ag}$  in the various kind of squid collected from around Japan during 1978-1989. As the results of analysis of squid, we have found that squid is useful material to investigate the radioactivity level of marine environment.

In this paper, we describe the analytical results on the trace elements in squids and their organs by inductively coupled plasma-mass spectrometry (ICP-MS) and instrumental neutron activation analysis (INAA) as well as the radioactive nuclides ( $^{60}\text{Co}$ ,  $^{108\text{m}/110\text{m}}\text{Ag}$  and  $^{137}\text{Cs}$ ) by  $\gamma$ -spectrometry.

## 2. EXPERIMENTAL

### 1) Sample collection and preparations

Three varieties of squids (*Todarodes pacificus*, *Loligo edulis*, *Sepia officinalis*) had been collected from the eight local fishery cooperative unions in Japan during 1981-1988, dried at 105°C and ashed for 24 hour at 450°C. Especially, *Todarodes pacificus* collected from Chiba fishery cooperative union in 1996 was divided into edible part, backbone, liver, stomach including diets, and other viscera.

### 2) Determination of trace elements in squids

#### (1) ICP-MS analysis

0.5 gram of squid ash samples was taken in the teflon beaker and 5 ml of 7 M  $\text{HNO}_3$  was added, heated to dryness. The residue was dissolved in 20 ml of 1 M  $\text{HNO}_3$  and boiled for 10 min. The sample solution was transferred to 100 ml teflon

Table 1. Isotopes and internal standards used for analysis of biological samples.

Element	Mass number of isotope used	Internal standard	Scan time (msec)	Element	Mass number of isotope used	Internal standard	Scan time (msec)
V	51	Sc	120	Mo	98	Y	120
Cr	53	Sc	120	Ag	107	Rh	80
Mn	55	Sc	120	Cd	111	In	80
Fe	57	Sc	120	Cs	133	In	80
Co	59	Sc	120	Ba	137	In	80
Ni	62	Sc	120	Tl	205	Tb	50
Cu	63	Sc	120	Pb	207	Tb	50
Zn	66	Sc	120	Bi	209	Tb	50
Se	77	Y	120	Th	232	Tb	50
Rb	85	Y	120	U	238	Tb	50
Sr	88	Y	120				

flask with 1 M HNO<sub>3</sub>. When required, samples were diluted 10 or 50 times with 18.2 megaohm pure water produced freshly by Milli-Ro60 (Millipore). The sample solutions were analyzed for V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Se, Rb, Sr, Mo, Ag, Cd, Cs, Ba, Tl, Pb, Bi, Th and U. The mass number of isotopes used for ICP-MS measurement are presented in table 1. 10 ng/ml of Sc, Y, Rh, In and Tb was separately added to the sample solutions as an internal standard.

The standard stock solution of 10 mg/l (SPEX multi-element plasma standard, SPEX chemical, USA) was diluted to prepare a series of standard solution (0, 0.05, 0.5, 2.5, 5, 50, 100, 200 ng/ml).

The data obtained from ICP-MS were calculated by spreadsheet software "Quattro Pro (Boland)" installed in personal computer.

## (2) INAA analysis

For the analysis of V, Mn and Cu, 100 mg of squid ash samples or the standards impregnated with filter paper were sealed in polyethylene bag, irradiated for 2 min by TRIGA Mark II reactor (100 kW, flux of thermal neutron;  $1.5 \times 10^{12} \text{ n cm}^{-2} \text{ sec}^{-1}$ ) of RIKKYO University in Japan. After the sample was cooled down for 6 min, radioactivities of <sup>52</sup>V, <sup>66</sup>Cu and <sup>56</sup>Mn were counted for 300 sec using the

multichannel analyzer with Ge detector with 19 and 35% relative efficiency for 1.332 MeV gamma rays of <sup>60</sup>Co. The measurement of <sup>56</sup>Mn (half life: 2.58h, 0.8438 meV) was made after <sup>27</sup>Mg (half life: 9.46m, 0.8468 meV), interfering <sup>56</sup>Mn counting had decayed out (about 1 hr) (Morrison and Potter, 1972).

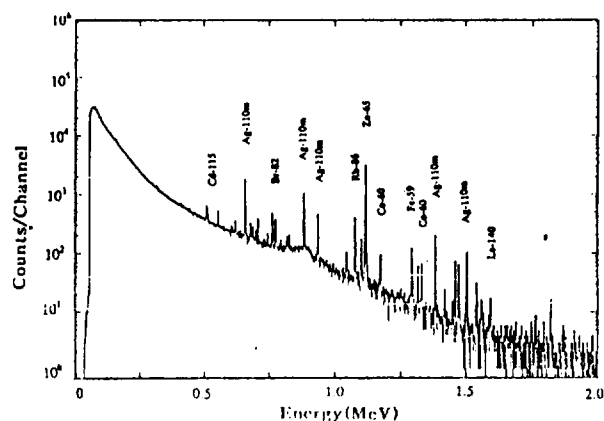


Figure 1.  $\gamma$ -ray spectrum of squid ash sample counted by Ge detector one month after neutron irradiation

For the analysis of Fe, Cs, Ag, Co and Zn 200 mg of samples and standards were irradiated for 12 hr at the flux of  $5 \times 10^{11} \text{ n cm}^{-2} \text{ sec}^{-1}$ . After the sample was cooled down for 2 weeks - 1 month, radioactivities of <sup>59</sup>Fe, <sup>60</sup>Co, <sup>65</sup>Zn, <sup>86</sup>Rb, <sup>110m</sup>Ag, <sup>115m</sup>In and <sup>115m</sup>Cd were counted for 600-4000 sec under the

conditions of plastic plate inserted between detector and sample to reduce intensity of the gamma rays induced from intense  $^{32}\text{P}$  in sample (Morrison and Potter, 1972). Figure 1 shows the  $\gamma$ -ray spectrum of squid ash sample counted by Ge detector 1 month after neutron irradiation.

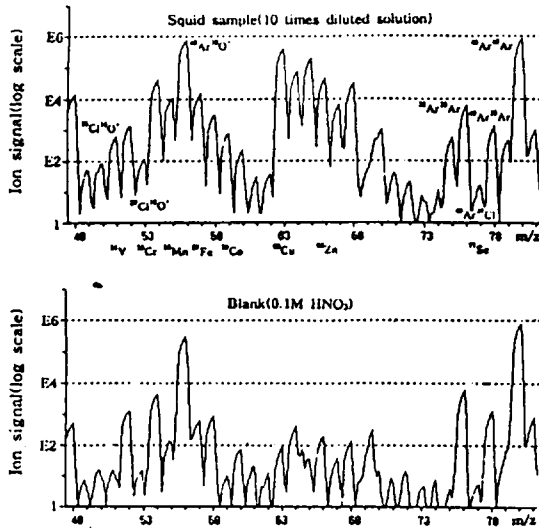


Figure 2. Mass spectrum of squid ash sample solution. After 0.1 g of squid ash sample was taken and dissolved in 20 ml of 1M  $\text{HNO}_3$  and boiled. The sample was transferred to 100 ml flask with 1M  $\text{HNO}_3$  and diluted 10 times.

### 3) $\gamma$ -spectrometry

After the ash samples were uniformly packed into a cylindrical vessel of 48 mm diameter, 60 mm height,  $^{60}\text{Co}$ ,  $^{108\text{m}/110\text{m}}\text{Ag}$  and  $^{137}\text{Cs}$  activities in squid ash sample were counted for about 80,000 sec by  $\gamma$ -spectrometry with Ge detector having approximately 20 % relative efficiency for 1.332 MeV gamma rays of  $^{60}\text{Co}$ . The counting efficiency of Ge detector is determined by using a mixed point source and KCl volume source with different height.

## 3. RESULT AND DISCUSSIONS

### 1) Elimination of spectral interferences of ICP-MS

Inductively coupled plasma-mass spectrometry (ICP-MS) is now a well-established analytical technique that provides with very low detection limits (1  $\mu\text{g/L}$  or less) for most of the elements in the periodic table. Also it allows multi-element analysis with rapid sample throughput (Dams *et al.*, 1995). This technique has, however, some difficulties associated with analytical performance such as matrix effect and molecular ion interference.

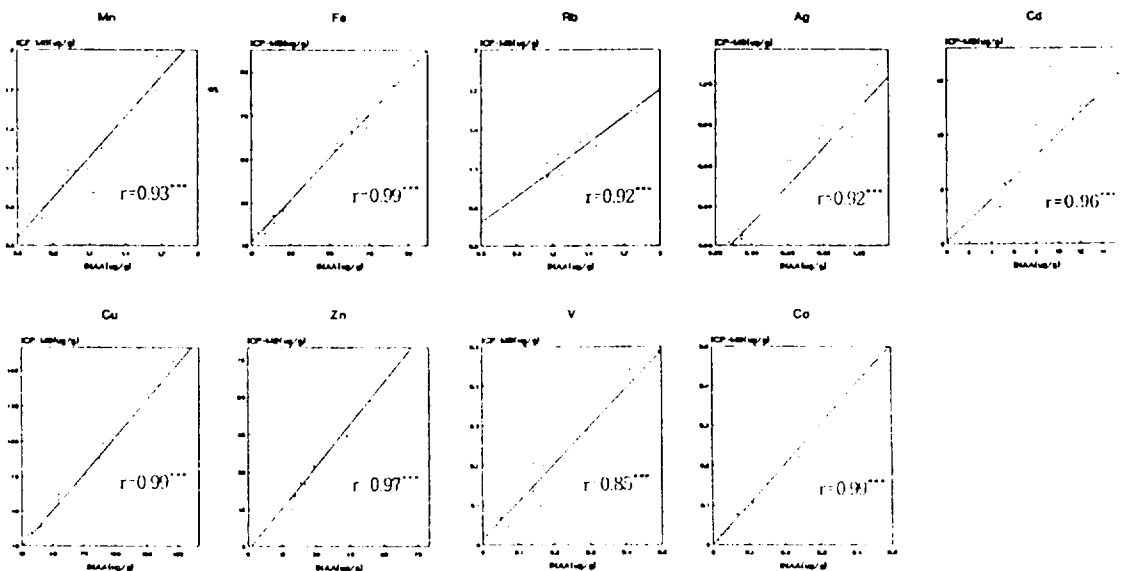


Figure 3. Comparison of ICP-MS and INAA results, as determined the concentrations of trace elements in squid by their techniques.

The problems of the background spectral interference (e.g.  $\text{ArO}^+$ ,  $\text{Ar}_2$  etc.) of blank sample can be solved to some extent by optimizing ICP-MS performance. However, the contribution of polyatomic ions and oxides from biological sample matrix on  $m/z$  peak of elements ( $^{35}\text{ClO}^+$ ,  $\text{ArC}^+$ ,  $^{37}\text{ClO}^+$ ,  $\text{ArO}^+$ ,  $\text{PO}_2$ ,  $\text{ArNa}^+$ ,  $^{35}\text{Cl}_2$  or  $\text{Ar}_2$  occurred as a result of combinations of C, Cl, S, O, P and Na) is the most severe problem for the determination of lighter elements ( $m/z < 80$ ) (Houk and Thompson, 1988; Dams *et al.*, 1995), giving higher background to reduce the accuracy and precision of analysis because the peaks of some polyatomic ions and oxides are overlapped with those from analytes. Therefore, in squid sample possible interferences may be caused by Na, K, Ca, Mg, Cl, S or P. The peaks of polyatomic ions and oxides generated from those matrixes were identified by introducing the solution containing Na, K, Ca, Mg, Cl, P or S (data not shown). The important spectral overlapping related to the determination of squid samples occurred on peaks of  $^{51}\text{V} (^{35}\text{Cl}^{16}\text{O}^+)$ ,  $^{53}\text{Cr} (^{37}\text{Cl}^{16}\text{O}^+)$ ,  $^{77}\text{Se} (^{40}\text{Ar}^{37}\text{Cl}^+)$  (figure 2), thus being limited to Cl effect. However, as the contribution of Cl on  $^{35}\text{Cl}^{16}\text{O}^+$ ,  $^{37}\text{Cl}^{16}\text{O}^+$  and  $^{40}\text{Ar}^{37}\text{Cl}^+$  at the different Cl concentrations (0 - 50 mg/l) was very small, those spectral interferences can be eliminated by diluting sample solution to be below 20 mg/l.

## 2) Analysis of trace elements

Under the optimum condition of ICP-MS measurement previously mentioned, the simultaneous multi-analysis of 21 elements (V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Se, Rb, Sr, Mo, Ag, Cd, Cs, Ba, Tl, Pb, Bi, Th, U) in the squid sample was tried. As there are no available certified values for trace elements in the squid, the method of INAA was applied to compare with the results of ICP-MS. In the practical measurement, besides spectral interference by polyatomic ions derived from matrix elements such as Na, Ca and Cl, the count rates of analytes

can be affected by high level of salts or heavy matrix ions themselves (Houk and Thompson, 1988). Orifice plugging is also, a problem for the samples with high solid contents (> 0.5 %). Hence, internal standard was adapted to ICP-MS to deal with matrix effects. Five elements (Sc, Y, In, Tb, Rh) well-known as internal standard were used.

Three biological reference samples (SRM 1572 citrus leaves, SRM 1577 bovine liver, NIES No. 9 sargasso) were analyzed to obtain the validity of our ICP-MS results, and in general our analytical data were well agreed with the certified values in all standard materials (data not shown). On the other hand, the concentrations of V, Mn, Fe, Co, Cu, Zn, Rb, Ag and Cd determined by ICP-MS were closely correlated with those of INAA as shown in figure 3.

The reason why the analytical result of Ag by ICP-MS is less than that by INAA could be explained from the fact that Ag is apt to absorb on teflon vial wall and its concentration can be decreased as time elapsed. This is one of the difficulties of Ag determination by ICP-MS.

The concentration of trace elements in squids was shown in table 2. The concentrations of Fe, Ni, Cu, Zn, Mo, Cd, Pb and U in *Todarodes pacificus*, a migratory squid, were higher than those in other varieties. *Sepia officinalis*, a sedentary one, had higher concentration of V, Co, Se, Sr. However, there were no differences among the varieties in the concentrations of other elements. The concentrations of trace elements in *Todarodes pacificus* were influenced by the sampling sites.

Based on these results, there might be physiologically different characteristics among squid varieties. It might be considered that the different Cd values in *Todarodes pacificus* among sampling sites represents some possibility of environmental contamination.

Table 2. The concentrations( $\mu\text{g/g}$ ) of trace elements in squids as determined by ICP-MS and INAA.

Location	Variety	Mn	Fe	Cu	Zn	Rb	Sr	Ag	V	Cr	Co	Ni	Se	Mo	Cd	Cs	Ba	Tl	Pb	Bi	Th	U
Hokkaido	<i>Todarodes pacificus</i>	0.96*	83.8	85.4	72.6	1.26	4.09	0.85	0.440	0.023	0.153	0.229	0.039	0.054	15.6	0.004	0.146	0.0001	0.339	0.0025	0.0004	0.0347
		0.88**	82.0	84.3	68.8	1.18			0.91	0.410		0.150				15.3				0.213	0.0028	0.0001
Hokkaido	//	0.71	68.7	54.6	57.8	1.03	2.63	0.69	0.065	0.017	0.104	0.077	0.027	0.012	5.77	0.003	0.013	0.0000	0.094	0.0007	0.0002	0.0615
		0.70	63.4	45.3	49.1	1.05			0.67	0.090	0.110				5.63				0.094	0.0007	0.0002	0.0615
Miyagi	//	1.07	77.1	131	43.9	1.52	5.22	0.86	0.488	0.037	0.193	0.393	0.074	0.017	13.0	0.005	0.073	0.0001	0.095	0.0014	0.0001	0.0167
		0.98	81.3	131	43.4	1.80	5.85	0.69	0.113	0.009	0.074	0.166	0.028	0.009	5.55	0.003	0.043	-	0.095	0.0014	0.0001	0.0167
Chiiba	//	0.60	53.2	44.8	28.6	0.83	10.9	0.42	0.205	0.021	0.132	0.643	0.076	0.026	10.8	0.004	0.198	0.0002	0.119	0.0022	0.0003	0.0222
		1.21	49.0	168	59.4	1.24			0.64	0.060	0.070	0.100	0.076	0.034	8.00				0.299	0.0016	0.0002	0.0158
Niigata	//	0.80	54.2	77.5	35.7	1.05	3.66	0.94	0.045	0.016	0.076	0.076	0.034	0.007	9.19	0.003	0.026	0.0001	0.292	0.0036	0.0002	0.0297
		0.72	52.3	79.7	33.0	1.06	4.93	1.39	0.135	0.016	0.107	0.179	0.013	0.011	11.6				0.292	0.0036	0.0002	0.0297
Niigata	//	0.96	60.6	97.4	51.5	1.23	3.46	1.41	0.140	0.029	0.086	0.129	0.027	0.012	12.0	0.003	0.038	0.0001	0.429	0.0036	0.0003	0.0255
		0.83	64.4	97.9	52.4	1.07			1.32	0.060	0.087	0.087			15.4							
Submean(ICP-MS)	CV(*)	0.91	64.7	96.1	50.9	1.14	5.09	0.88	0.194	0.021	0.116	0.237	0.040	0.019	11.0	0.003	0.072	0.0001	0.235	0.0023	0.0002	0.0287
		17.8	16.9	38.2	24.2	16.8	47.2	32.8	84.3	39.2	33.5	76.3	54.4	76.5	33.5	19.5	85.6	87.2	49.7	42.5	45.6	47.6
Submean(INAA)	CV(*)	0.81	62.9	84.5	45.6	1.19			1.02	0.181	0.111				10.8							
		15.9	20.9	32.1	25.2	22.0			25.9	88.0	37.5				35.4							
Kagoshima	<i>Loligo edulis</i>	1.04	20.3	18.6	31.8	1.02	13.4	0.09	0.044	0.008	0.032	0.048	0.124	0.005	0.24	0.008	0.084	-	0.012	0.0003	0.0001	0.0061
		0.91	21.3	15.5	28.7	1.00			0.27		0.030	0.030	0.033	0.040	0.006	0.18	0.006	0.106	-	0.009	0.0007	0.0007
Fukui	//	0.76	24.0	21.5	20.2	1.36	10.1	0.17	0.071	0.005	0.024	0.024	0.024	0.006	0.18	0.006	0.106	-	0.023	0.0000	0.0002	0.0056
		0.77	20.5	18.1	20.5	1.38	8.22	0.07	0.097	0.005	0.038	0.038	0.056	0.004	0.15	0.006	0.067	-	0.023	0.0000	0.0002	0.0056
Nagasaki	<i>Loligo edulis</i>	1.66	12.8	28.3	22.8	1.14	7.42	0.16	0.160	0.001	0.040	0.042	0.063	0.005	0.12	0.006	0.044	0.0003	0.033	0.0002	0.0001	0.0048
		1.95	17.7	28.7	25.1	1.30			0.05	0.062	0.001	0.030	0.042	0.063	0.005	0.12	0.006	0.044	0.0003	0.033	0.0002	0.0001
Kagoshima	//	1.66	16.0	34.0	25.0	1.33	7.50	0.19	0.030	0.003	0.030	0.047	0.116	0.005	0.25	0.005	0.038	0.0004	0.019	0.0003	0.0003	0.0066
		1.08	37.3	20.3	27.6	1.27			0.10	0.068	0.003	0.024	0.047	0.098					0.019	0.0003	0.0003	0.0052
Submean(ICP-MS)	CV(*)	1.30	23.0	23.1	26.1	1.23	9.33	0.26	0.050	0.004	0.029	0.042	0.080	0.005	0.19	0.006	0.068	0.0001	0.019	0.0003	0.0003	0.0052
		34.0	33.4	16.4	14.4	9.60	24.1	45.0	25.1	55.0	17.8	13.5	42.4	14.7	27.5	18.2	37.0	129	44.3	70.2	79.8	23.1
Submean(INAA)	CV(*)	1.18	22.1	23.1	24.7	1.26			0.22	0.052	0.029	0.029	0.030									
		33.1	4.24	30.0	11.5	13.3			18.9	108	22.9			125								
Okinawa	<i>Sepia officinalis</i>	1.25	27.7	71.7	58.2	1.09	23.2	0.29	0.199	0.027	0.387	0.173	0.223	0.019	3.47	0.004	0.026	0.0001	0.064	0.0012	0.0004	0.0122
		1.20	26.8	70.3	54.0	1.15			0.58	0.170	0.370				3.35				0.260	0.0003	0.0004	0.0118
Okinawa	//	1.09	26.0	25.1	44.6	1.05	27.2	0.12	0.286	0.022	0.220	0.116	0.179	0.014	4.77	0.004	0.092	0.0002	0.260	0.0003	0.0004	0.0118
		1.08	26.1	25.3	43.7	1.07			0.27	0.290	0.239				4.77				0.162	0.0007	0.0004	0.0120
Submean(ICP-MS)	CV(*)	1.17	26.8	48.4	51.4	1.07	25.2	0.21	0.243	0.025	0.304	0.144	0.201	0.017	3.41	0.004	0.059	0.0001	0.162	0.0007	0.0004	0.0120
		6.79	3.27	48.1	13.2	1.91	7.87	40.5	17.9	9.938	27.5	19.7	11.0	15.8	1.65	9.05	56.4	62.6	60.4	65.7	3.47	1.76
Submean(INAA)	CV(*)	1.14	26.5	47.8	48.9	1.16			0.43	0.23	0.305	0.305	0.030		4.06				0.153	0.0014	0.0003	0.0187
		5.26	1.32	47.1	10.5	0.86			36.5	26.1	21.5				17.5				0.153	0.0014	0.0003	0.0187
Mean(ICP-MS)	CV(*)	1.08	45.7	65.4	42.7	1.16	9.19	0.53	0.16	0.016	0.112	0.159	0.075	0.014	6.39	0.004	0.069	0.0001	0.153	0.0014	0.0003	0.0187
		31.1	48.6	67.2	35.5	14.2	75.7	82.1	86.8	65.0	84.3	99.9	79.1	88.7	89.4	34.0	71.1	113	87.8	84.4	58.5	79.5
Mean(INAA)	CV(*)	0.98	44.4	59.1	39.0	1.21			0.68	0.14	0.110											
		31.1	51.0	60.2	34.4	18.0			63.0	96.9	85.3											

\*: ICP-MS determination, \*\*: INAA determination, -: not detected

Table 3. The concentrations( $\mu\text{g/g}$ ) and enrichment factors of trace elements in *Todarodes pacificus* organs as determined by ICP-MS.

Description	Mn	Fe	Cu	Zn	Rb	Sr	Ag	V	Cr	Co	Ni	Se	Mo	Cd	Ce	Ba	Tl	Pb	Bi	Th	U
Edible part	C	0.22	6.30	3.55	13.9	1.28	1.67	0.002	0.004	0.035	0.007	0.082	0.030	0.33	0.004	0.039	0.0001	0.054	0.0000	0.0001	0.0013
	E	6687	1853	27339	53593	12	0.19	2246	2	106735	4023	165	270	0.176	5914	14	3339	10	3701	603	287
Back bone	C	0.84	9.91	28.4	152	1.17	3.31	0.001	0.088	0.090	0.048	0.171	0.275	3.75	0.004	0.106	0.0027	0.291	0.0000	0.0001	0.0581
	E	25510	2915	218743	582998	11	0.38	1272	44	271953	28926	342	2502	6.439	67000	13	9021	222	20091	0	181
Viscera*	C	0.90	9.26	54.5	31.2	1.55	4.50	0.077	0.033	0.047	0.038	0.069	0.030	0.30	0.012	0.053	0.0002	0.132	0.0005	0.0002	0.0078
	E	27388	2723	419113	119852	15	0.51	77044	16	143132	20920	139	273	0.431	5326	38	4502	17	9128	6660	325
Liver	C	1.17	161	101	68.3	1.23	2.53	1.85	0.920	0.080	0.510	0.470	0.310	39.49	0.004	0.078	0.0006	2.150	0.0020	0.0004	0.0027
	E	35455	47474	779923	262692	12	0.29	1850000	460	242424	283333	940	2818	1.875	705179	13	6667	49	148276	27397	800
Others**	C	5.17	92.3	20.6	49.5	1.22	40.7	0.058	0.317	0.112	0.034	0.118	0.090	1.28	0.006	0.967	0.0002	0.152	0.0006	0.0013	0.0202
	E	156667	27159	158653	190278	12	4.62	58269	159	338635	18906	236	816	0.708	22821	22	82634	19	10490	7940	2520

\*: not including liver, and stomach with diets \*\*: stomach including diets C: concentration E: enrichment factor

### 3) Distributions of trace elements in squid organ and concentration factor

Examination of tissue-specific distributions of trace elements in fish organ and concentration factor would be extremely informative in estimating the environmental contamination as well as artificial radionuclide's behavior in marine ecosystem. *Todarodes pacificus* collected in 1996 was divided into edible part, backbone, liver, stomach including diets and other viscera, and trace elements in each organ were analyzed using ICP-MS and concentration factors for all metals were calculated from the concentrations of those in sea water(Nozaki, 1987)(table 3).

It is well known that liver is a very important organ in living organisms and have lots of metallo-enzymes. The concentrations and concentration factors of Fe(161  $\mu\text{g/g}$ ,  $4.8 \times 10^4$ ) Cu(101  $\mu\text{g/g}$ ,  $7.8 \times 10^5$ ), Zn(68.3  $\mu\text{g/g}$ ,  $2.6 \times 10^5$ ), V(0.92  $\mu\text{g/g}$ , 460), Co(0.51  $\mu\text{g/g}$ ,  $2.8 \times 10^5$ ) and Se(0.31  $\mu\text{g/g}$ ,  $2.8 \times 10^3$ ) being very essential for enzyme functions in liver were higher than in other organs, which were consistent with the report of Ueda *et al.*(1979). In addition, the concentration of Cd(39.5  $\mu\text{g/g}$ ,  $7.0 \times 10^5$ ), Ag(1.85  $\mu\text{g/g}$ ,  $1.85 \times 10^6$ ), Pb(2.15  $\mu\text{g/g}$ ,  $1.4 \times 10^5$ ) and Bi(2 ng/g,  $2.7 \times 10^4$ ) in liver were highly

accumulated and thus, very high concentration factors were calculated. On the other hand, the concentrations and concentration factors of Mn(stomach and diets: 2.15  $\mu\text{g/g}$ ,  $1.4 \times 10^5$ ), Cr(0.112  $\mu\text{g/g}$ ,  $3.4 \times 10^5$ ) and Ba(0.97  $\mu\text{g/g}$ ,  $8.3 \times 10^4$ ) in the viscera not including liver also were rather high. Therefore, it is suggested that the characteristic distribution patterns of trace elements in squid organs offer the possibility of its application to study of marine environments and squid viscera can be a good indicator organ in estimating the marine environment.

### 4) $^{60}\text{Co}$ , $^{108\text{m}/110\text{m}}\text{Ag}$ , and $^{137}\text{Cs}$ determination in squids

Since the artificial radionuclides  $^{60}\text{Co}$ ,  $^{108\text{m}/110\text{m}}\text{Ag}$  and  $^{137}\text{Cs}$  are important indicators of radioactive pollution in marine ecosystem(Folsom and Young, 1965; Folsom *et al.*, 1970; Fukatsu *et al.*, 1982; Fukatsu and Higuchi, 1985 and 1987; Umezu, 1992), the concentrations of  $^{60}\text{Co}$ ,  $^{108\text{m}/110\text{m}}\text{Ag}$  and  $^{137}\text{Cs}$  in three varieties of squids were measured. As shown in table 4, 5, 6 and 7, radioactivities of  $^{60}\text{Co}$ ,  $^{108\text{m}/110\text{m}}\text{Ag}$  and  $^{137}\text{Cs}$  depending on species and sampling location were different. Interestingly, the level of  $^{108\text{m}}\text{Ag}$  activities in squids collected after

Table 4. Stable Co concentration,  $^{60}\text{Co}$  activity and Co specific activity in squids taken from Japan near-seas.

Location	Collection date	Variety	$^{60}\text{Co}$ (mBq/kg wet)		Stable Co (mg/kg wet)	Specific activity (Bq/g Ag)
			A*	B**		
Hokkaido	Nov. 10, 86	<i>Todarodes pacificus</i>	18.9	5.67	0.141	40
Hokkaido	Nov. 11, 88	<i>Todarodes pacificus</i>	12.6	4.94	0.097	51
Miyagi	Sept. 19, 88	<i>Todarodes pacificus</i>	11.8	4.54	0.200	23
Chiba	June 30, 86	<i>Todarodes pacificus</i>	24.4	7.00	0.065	108
Chiba	July 3, 88	<i>Todarodes pacificus</i>	29.2	10.9	0.105	104
Niigata	July 13, 81	<i>Todarodes pacificus</i>	51.1	7.62	0.065	117
Niigata	June 29, 88	<i>Todarodes pacificus</i>	25.2	7.21	0.108	67
Niigata	June 27, 88	<i>Todarodes pacificus</i>	13.0	4.82	0.087	56
Fukui	Oct. 9, 86	<i>Loligo edulis</i>	7.4	2.20	0.018	121
Okinawa	July 4, 86	<i>Sepia officinalis</i>	40.7	11.7	0.346	34
Okinawa	July 19, 88	<i>Sepia officinalis</i>	47.0	17.6	0.239	74

\*: Activity at sampling date

\*\* : Activity decav-corrected to Jan. 1, 1996

Table 5. Stable Ag concentration,  $^{110m}\text{Ag}$  activity and Ag specific activity in squids taken from Japan near-seas.

Location	Collection date	Variety	$^{110m}\text{Ag}$ (mBq/kg wet)		Stable Ag (mg/kg wet)	Specific activity (Bq/g Ag)
			A*	B**		
Hokkaido	Nov. 10, 86	<i>Todarodes pacificus</i>	10,730	1.101	0.91	1.21
Hokkaido	Nov. 11. 88	<i>Todarodes pacificus</i>	70.3	0.056	0.67	0.08
Miyagi	Sept. 19, 88	<i>Todarodes pacificus</i>	181	0.120	1.24	0.10
Chiba	June 30, 86	<i>Todarodes pacificus</i>	4,070	0.290	0.64	0.45
Chiba	July 3, 88	<i>Todarodes pacificus</i>	218	0.117	1.10	0.11
Niigata	July 13, 81	<i>Todarodes pacificus</i>	185	8.990	0.88	0.00
Niigata	June 29, 88	<i>Todarodes pacificus</i>	9,620	0.683	1.41	0.49
Niigata	June 27, 88	<i>Todarodes pacificus</i>	185	0.098	1.39	0.07
Kagoshima	June 22, 86	<i>Loligo edulis</i>	229	0.016	0.22	0.07
Fukui	Oct. 9, 86	<i>Loligo edulis</i>	1,147	0.108	0.24	0.46
Nagasaki	June 18, 88	<i>Loligo edulis</i>	340	0.023	0.17	0.14
Okinawa	July 4, 86	<i>Sepia officinalis</i>	1,295	0.093	0.58	0.16

\*: Activity at sampling date

\*\*: Activity decay-corrected to Jan. 1, 1996

Table 6. Stable Ag concentration,  $^{108}\text{mAg}$  activity and Ag specific activity in squids taken from Japan near-seas.

Location	Collection date	Species	$^{108m}\text{Ag}$ (mBq/kg wet)		Stable Ag (mg/kg wet)	Specific activity (Bq/g Ag)
			A*	B**		
Hokkaido	Nov. 11. 88	<i>Todarodes pacificus</i>	34.0	32.7	0.67	49
Chiba	July 3, 88	<i>Todarodes pacificus</i>	74	71.0	1.10	65
Niigata	July 13, 81	<i>Todarodes pacificus</i>	44.4	41.0	0.88	47
Kagoshima	June 22, 86	<i>Loligo edulis</i>	229	217	0.22	990
Nagasaki	June. 18, 88	<i>Loligo edulis</i>	340	323	0.17	1,901
Okinawa	July 4, 86	<i>Sepia officinalis</i>	74	70.3	0.58	120

\*: Activity at sampling date

\*\*: Activity decay-corrected to Jan. 1, 1996

Table 7. Stable Cs concentration,  $^{137}\text{Cs}$  activity and Cs specific activity in squids taken from Japan near-seas.

Location	Collection date	Variety	$^{137}\text{Cs}$ (mBq/kg wet)		Stable Cs (mg/kg wet)	Specific activity (Bq/g Ag)
			A*	B**		
Hokkaido	Nov. 10, 86	<i>Todarodes pacificus</i>	66.6	54.0	0.004	13,496
Hokkaido	Nov. 11. 88	<i>Todarodes pacificus</i>	32.2	27.3	0.003	9,115
Miyagi	Sept. 19, 88	<i>Todarodes pacificus</i>	111	93.9	0.005	18,780
Niigata	July 13, 81	<i>Todarodes pacificus</i>	62.9	45.1	0.003	15,037
Okinawa	July 4, 86	<i>Sepia officinalis</i>	88.8	71.4	0.004	17,849

\*: Activity at sampling date

\*\*: Activity decay-corrected to Jan. 1, 1996



Chernobyl accident in April 1986 were increased. The accumulation was much more in *Todarodes pacificus*, a migratory squid than in *Loligo edulis* and *Sepia officinalis*, sedentary ones. However,  $^{110m}\text{Ag}$  (127.7 y) activities having longer physical half life were higher in *Loligo edulis* than other squids. It seemed to be closely related to regional level of  $^{110m}\text{Ag}$  in marine environment.

Unlike  $^{108m/110m}\text{Ag}$ ,  $^{60}\text{Co}$  levels in squids were generally decreased year by year. *Sepia officinalis* has higher  $^{60}\text{Co}$  activity than *Todarodes pacificus* and *Loligo edulis*. On the other hand,  $^{137}\text{Cs}$  specific activities in squids were 10 - 100,000 times higher than those of  $^{60}\text{Co}$  or  $^{108m/110m}\text{Ag}$ .

#### 4. CONCLUSION

This study shows that the inductively coupled plasma-mass spectrometry (ICP-MS) can be used for the simultaneous determination of 21 trace elements in squid. In the practical ICP-MS measurement of squid ash samples, corrections for spectral interferences caused by polyatomic ions of  $\text{Cl}(^{35}\text{Cl}^{16}\text{O}^+$  on  $^{51}\text{V}$  and  $^{40}\text{Ar}^{37}\text{Cl}^+$  on  $^{77}\text{Se}$ ) were not required. Since there was a good relation between ICP-MS and INAA in the determination of V, Mn, Fe, Co, Cu, Zn, Rb, Ag and Cd, ICP-MS is thought to be a reliable technique for the trace element analysis of squids.

The different chemical composition between two varieties was observed: high concentrations of Fe, Cu, Zn, Ni, Mo, Cd, Pb and U in *Todarodes pacificus*, a migratory squid and of Sr, V, Co, and Se in *Sepia officinalis*, a sedentary one. The change of Cd values in squid samples from different sampling sites might be caused due to the possible contamination of their environment.

The analytical data for the stable isotopes measured by ICP-MS were closely correlated with the results of radioactive counting of  $^{60}\text{Co}$ ,  $^{110m}\text{Ag}$  and  $^{137}\text{Cs}$  by gamma spectrometry.

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